2.20 Pikes Peak near Denver, Colorado has an elevation of 14,110 ft. (a) Determine the pressure at this elevation, based on Eq. 2.12. (b) If the air is assumed to have a constant specific weight of 0.07647 lb/ft³, what would the pressure be at this altitude? (c) If the air is assumed to have a constant temperature of 59 °F what would the pressure be at this elevation? For all three cases assume standard atmospheric conditions at sea level (see Table 2.1).

2.20

(a)
$$p = p_a \left(1 - \frac{B^2}{T_a}\right)^{\frac{2}{R_B}}$$
 (Eq. 2.12)

For
$$P_a = 2116.2 \frac{lb}{ft^2}$$
, $B = 0.00357 \frac{oR}{ft}$, $g = 32.174 \frac{ft}{s^2}$,

 $T_a = 518.67 \, ^{\circ}R$, $R = 1716 \frac{ft \cdot lb}{slug \cdot ^{\circ}R}$, and

$$\frac{g}{RB} = \frac{32.174 \frac{ft}{s^2}}{\left(1716 \frac{ft \cdot lb}{slug \cdot ^{\circ}R}\right) \left(0.00357 \frac{oR}{ft}\right)} = 5.252$$

Then
$$p = \left(2116.2 \frac{1b}{ft^2}\right) \left[1 - \frac{\left(0.00357 \frac{OR}{ft}\right) \left(14,110 ft\right)}{518.67 \, ^{\circ}R}\right]$$

$$= 1240 \frac{16}{ft^2} (abs)$$

(6)
$$p = p_a - 8h$$

$$= 2116.2 \frac{15}{ft^2} - (0.67647 \frac{16}{ft^3})(14, 110 ft)$$

$$= \frac{1040 \frac{16}{ft^{2}} (abs)}{-\frac{gh}{RTa}}$$
(C)
$$p = p_{a} e^{\frac{1}{RTa}}$$
(Eq. 2.10)

$$= (2116.2 \frac{16}{ft^2}) e^{-\left(\frac{32.174 \frac{fb}{5^2}}{(1716 \frac{ft\cdot 16}{5lug\cdot 0R})(518.670R)}\right)}$$

$$= 1270 \frac{16}{ft^2} (a6s)$$