

2.21

2.21 Pikes Peak near Denver, Colorado has an elevation of 14,110 ft. (a) Determine the pressure at this elevation, based on Eq. 2.12. (b) If the air is assumed to have a constant specific weight of 0.07647 lb/ft^3 , what would the pressure be at this altitude? (c) If the air is assumed to have a constant temperature of 59°F what would the pressure be at this elevation? For all three cases assume standard atmospheric conditions at sea level (see Table 2.1).

$$(a) \quad p = p_a \left(1 - \frac{\beta z}{T_a} \right)^{\frac{g}{R\beta}} \quad (\text{Eq. 2.12})$$

$$\text{For } p_a = 2116.2 \frac{\text{lb}}{\text{ft}^2}, \quad \beta = 0.00357 \frac{^\circ\text{R}}{\text{ft}}, \quad g = 32.174 \frac{\text{ft}}{\text{s}^2}, \\ T_a = 518.67^\circ\text{R}, \quad R = 1716 \frac{\text{ft} \cdot \text{lb}}{\text{slug} \cdot ^\circ\text{R}}, \quad \text{and} \\ \frac{g}{R\beta} = \frac{32.174 \frac{\text{ft}}{\text{s}^2}}{(1716 \frac{\text{ft} \cdot \text{lb}}{\text{slug} \cdot ^\circ\text{R}})(0.00357 \frac{^\circ\text{R}}{\text{ft}})} = 5.252$$

then

$$p = (2116.2 \frac{\text{lb}}{\text{ft}^2}) \left[1 - \frac{(0.00357 \frac{^\circ\text{R}}{\text{ft}})(14,110 \text{ ft})}{518.67^\circ\text{R}} \right]^{5.252} \\ = \underline{\underline{1240 \frac{\text{lb}}{\text{ft}^2} \text{ (abs)}}}$$

$$(b) \quad p = p_a - \gamma h \\ = 2116.2 \frac{\text{lb}}{\text{ft}^2} - (0.07647 \frac{\text{lb}}{\text{ft}^3})(14,110 \text{ ft}) \\ = \underline{\underline{1040 \frac{\text{lb}}{\text{ft}^2} \text{ (abs)}}}$$

$$(c) \quad p = p_a e^{-\frac{gh}{RT_a}} \quad (\text{Eq. 2.10}) \\ = (2116.2 \frac{\text{lb}}{\text{ft}^2}) e^{-\frac{(32.174 \frac{\text{ft}}{\text{s}^2})(14,110 \text{ ft})}{(1716 \frac{\text{ft} \cdot \text{lb}}{\text{slug} \cdot ^\circ\text{R}})(518.67^\circ\text{R})}} \\ = \underline{\underline{1270 \frac{\text{lb}}{\text{ft}^2} \text{ (abs)}}}$$