

2.16

2.16 Pikes Peak near Denver, Colorado has an elevation of 14,110 ft. (a) Determine the pressure at this elevation, based on Eq. 2.12. (b) If the air is assumed to have a constant specific weight of 0.07647 lb/ft^3 , what would the pressure be at this altitude? (c) If the air is assumed to have a constant temperature of 59°F what would the pressure be at this elevation? For all three cases assume standard atmospheric conditions at sea level (see Table 2.1).

$$(a) \quad p = p_a \left(1 - \frac{\beta z}{T_a} \right)^{\frac{g}{R\beta}} \quad (\text{Eq. 2.12})$$

$$\text{For } p_a = 2116.2 \frac{\text{lb}}{\text{ft}^2}, \quad \beta = 0.00357 \frac{\text{OR}}{\text{ft}}, \quad g = 32.174 \frac{\text{ft}}{\text{s}^2},$$

$$T_a = 518.67^\circ\text{R}, \quad R = 1716 \frac{\text{ft}\cdot\text{lb}}{\text{slug}\cdot\text{OR}}, \quad \text{and}$$

$$\frac{g}{R\beta} = \frac{32.174 \frac{\text{ft}}{\text{s}^2}}{\left(1716 \frac{\text{ft}\cdot\text{lb}}{\text{slug}\cdot\text{OR}} \right) \left(0.00357 \frac{\text{OR}}{\text{ft}} \right)} = 5.252$$

then

$$p = \left(2116.2 \frac{\text{lb}}{\text{ft}^2} \right) \left[1 - \frac{(0.00357 \frac{\text{OR}}{\text{ft}})(14,110 \text{ ft})}{518.67^\circ\text{R}} \right]^{5.252}$$

$$= \underline{\underline{1240 \frac{\text{lb}}{\text{ft}^2} \text{ (abs)}}}$$

$$(b) \quad p = p_a - \gamma h$$

$$= 2116.2 \frac{\text{lb}}{\text{ft}^2} - \left(0.07647 \frac{\text{lb}}{\text{ft}^3} \right) (14,110 \text{ ft})$$

$$= \underline{\underline{1040 \frac{\text{lb}}{\text{ft}^2} \text{ (abs)}}}$$

$$(c) \quad p = p_a e^{-\frac{g h}{R T_a}} \quad (\text{Eq. 2.10})$$

$$= \left(2116.2 \frac{\text{lb}}{\text{ft}^2} \right) e^{-\left[\frac{(32.174 \frac{\text{ft}}{\text{s}^2})(14,110 \text{ ft})}{\left(1716 \frac{\text{ft}\cdot\text{lb}}{\text{slug}\cdot\text{OR}} \right) (518.67^\circ\text{R})} \right]}$$

$$= \underline{\underline{1270 \frac{\text{lb}}{\text{ft}^2} \text{ (abs)}}}$$