1. **Time and Place of Course**
   Mon / Wed 3:30 – 4:45  2133 SC

2. **Instructor**
   Charles (Charlie) Stanier
   Office: 2133 Seamans Center
   Phone: 335-1399
   Email: cstanier@engineering.uiowa.edu
   Office Hours: by appointment

   TA: Robert Bullard
   Office: 4th Floor of IATL
   Email: Robert-bullard@uiowa.edu
   Office Hours: by appointment

3. **Textbooks**

   Required

   Optional / Supplemental
   - Bohren, Craig, Clouds in a Glass of Beer. Dover Publications (July 10, 2001)
4. Target Audience & Prerequisites
   a) The class is aimed at a wide range of student levels – from seniors and beginning graduate students to advance graduate students.
   b) Assignments and projects will be tailored to the student’s level. Graduate students will be expected to do more comprehensive and fundamental projects. Undergraduates can use models as “black boxes” without exploring their inner workings. Problem sets may have two levels of questions (grad vs. undergrad) or questions that are optional for undergraduate students but required for graduate students.
   c) A course in reaction kinetics is required as a co-requisite. The ideal course for this is 52:105 Chemical Reaction Engineering or an equivalent course. Contact instructor for permission to waive this co-requisite requirement.
   d) Exposure to environmental chemistry, environmental science, aerosol technology, meteorology, numerical methods, computer programming, and/or environmental engineering is helpful, but not required.

5. Course Goals
   a) The primary goal of the class is to strengthen student knowledge of the fundamental and applied issues in atmospheric chemistry, through a combination of lectures, problem sets, and projects.
   b) By the end of this class, students should be able to follow (and place in scientific context) current research in atmospheric chemistry and physics – including field, laboratory, and computational research.
   c) The course gives exposure to research grade atmospheric simulations in selected problem sets and projects.
   d) Some topics of atmospheric chemistry will be covered at the survey & problem set level, including global circulation, global biogeochemical cycles, synoptic meteorology, vertical transport of pollutants, sampling techniques for gas phase compounds, aqueous phase reactions, deposition, gas-particle partitioning, photolysis, and atmospheric residence time.
   e) The class will focus in depth on one aspect of tropospheric chemistry: the HOx cycle and photochemical smog and ozone formation.
   f) Some topics of atmospheric aerosols at the survey & problem set level, including aerosol size distributions, dynamics of aerosol particles, dynamics of aerosol populations, radiative transfer involving aerosols, cloud formation, aerosol thermodynamics, and sampling techniques for aerosols.
   g) Time permitting, we will develop the basic physics of global climate change, including the issues of radiative balance, carbon cycle, and recent IPCC summary reports.
   h) Dispersion modeling and Chemical Transport Modeling by Eulerian and Langrangian methods will be discussed.
   i) Students will complete 1-2 projects on topics of their choice.
   j) In-depth assignments and mini-projects will be tailored to the students level. Graduate students will be expected to do more comprehensive and fundamental projects. Undergraduates can use models as “black boxes” without exploring their inner workings.

6. Course Outline
<table>
<thead>
<tr>
<th>Class</th>
<th>Date</th>
<th>Topic</th>
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<tbody>
<tr>
<td>1</td>
<td>Wed, Jan 18</td>
<td>Atmospheric Structure &amp; Constituents</td>
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<td>2</td>
<td>Mon, Jan 23</td>
<td>Atmospheric Structure &amp; Constituents</td>
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<tr>
<td>3</td>
<td>Wed, Jan 25</td>
<td>General Circulation / Horizontal Atmospheric Motion / Synoptic Meteorology</td>
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<td>4</td>
<td>Mon, Jan 30</td>
<td>Global Biogeochemical Cycles</td>
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<td>5</td>
<td>Wed, Feb 01</td>
<td>Worked Examples &amp; Discussion</td>
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<td>6</td>
<td>Mon, Feb 06</td>
<td>Chemical Kinetics of Atmos. Chem.</td>
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<td>7</td>
<td>Wed, Feb 08</td>
<td>Atmospheric Radiation &amp; Photochemistry</td>
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<td>8</td>
<td>Mon, Feb 13</td>
<td>Application of Atmospheric Chemical Kinetics: Stratospheric Chemistry, Chapman Cycle, H0x Budget</td>
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<td>9</td>
<td>Wed, Feb 15</td>
<td>Worked Examples &amp; Discussion</td>
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<td>10</td>
<td>Mon, Feb 20</td>
<td>Application of Atmospheric Chemical Kinetics: Stratospheric Ozone Depletion</td>
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<td>11</td>
<td>Wed, Feb 22</td>
<td>Tropospheric Chemistry I: OH and CO</td>
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<td>12</td>
<td>Mon, Feb 27</td>
<td>Tropospheric Chemistry II: HOx Cycle</td>
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<td>13</td>
<td>Wed, Feb 29</td>
<td>Tropospheric Chemistry III: Hydrocarbon Oxidation, NOx and VOC</td>
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<td>14</td>
<td>Mon, Mar 05</td>
<td>class cancelled - Stanier travel</td>
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<tr>
<td>15</td>
<td>Wed, Mar 07</td>
<td>Overview of Gas Measurement Techniques</td>
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<tr>
<td>Class</td>
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<td>SPRING BREAK (Mar 12-16)</td>
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<tr>
<td>16</td>
<td>Mon, Mar 19</td>
<td>Aqueous Phase Kinetics and Thermodynamics, S(IV) to S(VI), Acid Rain, and Haze</td>
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<tr>
<td>17</td>
<td>Wed, Mar 21</td>
<td>Aqueous Phase Kinetics and Thermodynamics, S(IV) to S(VI), Acid Rain, and Haze</td>
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<td>18</td>
<td>Mon, Mar 26</td>
<td>Overview of Atmospheric Aerosols</td>
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<td>19</td>
<td>Wed, Mar 28</td>
<td>Group Project Presentations</td>
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<td>20</td>
<td>Mon, Apr 02</td>
<td>Atmospheric aerosol cont.</td>
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<td>21</td>
<td>Wed, Apr 04</td>
<td>Aerosol Thermodynamics</td>
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<td>22</td>
<td>Mon, Apr 09</td>
<td>Aerosol Thermodynamics</td>
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<td>23</td>
<td>Wed, Apr 11</td>
<td>Vertical Stability, Local Scale Meteorology</td>
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<td>24</td>
<td>Mon, Apr 16</td>
<td>Guest Lecture: 3D Atmospheric Modeling</td>
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<td>25</td>
<td>Wed, Apr 18</td>
<td>Guest Lecture: 3D Atmospheric Modeling</td>
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<td>26</td>
<td>Mon, Apr 23</td>
<td>Overview of Aerosol Sampling Techniques</td>
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<td>27</td>
<td>Wed, Apr 25</td>
<td>Air Quality and Health - Case Study on U.S. Criteria Pollutants and Modeling to Achieve Compliance</td>
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<td>28</td>
<td>Mon, Apr 30</td>
<td>Individual Project Presentations</td>
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<td>29</td>
<td>Wed, May 02</td>
<td>Individual Project Presentations</td>
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Assignments

Homework assignments (~6 in number)
Group projects (1 in number)
Individual projects / presentations (1 in number)

Note: grading checklists will be provided for group and individual projects. These will be used by the instructor for grading and for peer assessment of oral presentations. Students will be asked to prepare questions for discussion sessions and will be asked to grade their own participation.

7. Quizzes
There may be 1-3 announced class quizzes that will have computations and be 30-60 min in length
There will be 2-4 unannounced quizzes that will be qualitative, be based on the reading, and be 5-10 min in length.

9. Exams
One take home exam.

NOTE ON EXAMS: Regrading of quizzes and exams will only be done if the exam was in ink. Quizzes and exams completed in pencil will not be rescored.

10. Grading
Portions of the class will be weighted using the following percentages
Problem Sets: 30%
Group Project 17.5%
Quizzes: 12.5%
Individual Project: 20%
Participation/attendance: 7.5%
Take home exam: 12.5%

Expectations:

A
• Completion of all or nearly all homework with attention to detail and honest effort
• “Pulling one’s weight” on group project with evidence of good teamwork; completion of group project with attention to detail, honest effort, neat & logical writeup, and plausible physical results
• Meeting of all milestone dates in the individual project; completion of individual project with attention to detail, honest effort, neat & logical writeup, and plausible physical results
• Demonstration in quizzes and take home exam that a majority of the material has been comprehended, that most of the basic computations and problem solving techniques discussed in the class can be done by the student, that all readings have been done, that appropriate units and physical plausibility of answers is discussed.
• High quality presentations and supporting analysis/calculations for projects OR significant improvement in presentations throughout semester. Specifically – want to see good slides, understanding of chemical and physical fundamentals, good delivery, and interest in subject material
• 100% attendance unless discussed prior to class with instructor
• Active participation in class

B

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- Severe deficiency in one area above
- Major deficiency in two areas above
- Minor deficiency in three or more areas above

C
- Severe deficiency in two or more areas above
- Major deficiency in three or more areas above
- Major deficiency in two areas above, and minor deficiency in more than two additional areas

11. Guidelines on Academic Misconduct and Other Matters

- Cheating on hour or final exams will result in an F in the course. Examples of cheating include but are not limited to looking at your neighbor’s exam papers, discussing problems during an exam, or copying answers from another exam paper. For a take home exam, communicating (verbally, nonverbally, or electronically) about what websites or textbooks may have useful information is also considered cheating.
- In this course, academic misconduct on the final individual project will result in an F for the course.
- Cheating on a quiz will result in a zero for the quiz portion of the class (e.g. a zero for all quizzes in the class). Examples of cheating include but are not limited to looking at your neighbor’s quiz papers, discussing problems during a quiz, or copying answers from another exam paper.
- Plagiarism on a lab report or project may result in penalties up to 0 in the quiz portion of the class (e.g. a zero for all lab reports) for all members in the lab group. Lesser penalties may be used at the instructor discretion.
- Acceptable collaboration on homework is defined as working on problems together. However, each student should write out the final calculation for themselves, and calculate any quantities using their own calculator or spreadsheet. This includes the visual basic simulator – while working together to get the basic idea of a VB simulator problem, the final parameter adjustments and graphs should be done individually. Otherwise, the person sitting at the computer may learn the most.
- Acceptible collaboration when using computer tools is similar. It is extremely useful to sit at neighboring computers and to talk to one another about lines of code, syntax, errors, and results, each student should create, run, edit, and comment their own code on their own college computing account. Working on a matlab program “together” (with one person typing and another looking over their shoulder) and then turning in two copies of the same program is not acceptable because the learning and preparation for quizzes and exams is not equal.

12. Policy statements:

12a. Disabilities:

Accommodations for Disabilities. If you feel that you may need an accommodation based on the impact of a disability please contact Prof. Charles Stanier privately to discuss your specific needs. You may also contact the Office of Student Disability Services (319/335-1462) to discuss the accommodations that are available for students with documented disabilities.

12b. Determining which college’s rules and regulations apply
This course is given by the College of Engineering but may be taken by students enrolled in colleges such as the Graduate College and the College of Public Health. The following document describes the applicability of policies from Engineering versus those from other colleges. Details of the University policy of cross enrollments may be found at: http://www.uiowa.edu/~provost/deos/crossenroll.doc).

12c. Attendance and federal financial aid.

Short version. Non-attendance of class, especially in the first 10 days, can lead to requirements for prompt repayment of federal financial aid.

Long version, from the office of the Provost:
Federal financial aid regulations require that The University of Iowa take additional steps during the semester to verify that students receiving federal financial aid are active in their courses. We want to stress that The University of Iowa does not have an attendance policy and we do not anticipate creating an attendance policy. However, to keep our students eligible to receive federal Title IV financial aid, the University must comply with the new regulations. We need your help to do so. Specifically, we are asking that all faculty and instructors respond to requests from the UI Registrar and UI Office of Student Financial Aid for specific information about the academic activity of their students.

The UI Office of Student Financial Aid may be able to reduce the financial liability of individual students by thousands of dollars based on your verification of academic activity (as defined below). During the Spring 2011 review, students for whom academic activity could not be verified were required to repay an average of $4700 in federal financial aid. For many students, this debt may be insurmountable and put an end to their academic careers. Your assistance is extremely important.

Each semester and for each class, the UI Registrar asks faculty and instructors to verify after the tenth day of the semester that a student attended class or participated in an academic activity (as defined by the U.S. Department of Education) on any day since the beginning of the class. The University REQUIRES 100% completion of the tenth day attendance report by faculty and instructors in order to comply with federal financial aid regulations. The UI Registrar also asks faculty and instructors to verify academic activity at midterm each semester. We request that all faculty and instructors complete these verification forms for every class.

If a student receiving federal financial aid receives ALL grades of I, F, U or N, during any semester, the UI Office of Student Financial Aid may contact you after the end of the semester to obtain a last date of academic activity. This date will determine whether or not a student has to repay a portion of his or her federal financial aid (grants and loans). The UI Office of Student Financial Aid is required by federal law to complete this review and adjust financial aid within 30 days of the end of the semester. We request that all faculty and instructors watch for an email of this nature and respond by the deadline provided in the email.